

Effect of SiO₂ and Al₂O₃ nanoparticles on the aggregative stability, structure formation, and strength of cementitious materials

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ABSTRACT

Introduction. The use of SiO₂ and Al₂O₃ nanoparticles opens new possibilities for controlling the structure of cementitious materials. However, their influence on the aggregative stability of the coagulation structure and on the strength, characteristics has been insufficiently studied. This study established the effect of SiO₂ and Al₂O₃ nanoparticles on the relationship between aggregative stability, early-age structure formation kinetics, and the strength of cementitious materials. **Materials and Methods.** Three cementitious systems based on Portland cement CEM I 42.5 were investigated: a reference system with a superplasticizer (Sika®ViscoCrete® 20 HE), as well as systems with the addition of SiO₂ nanoparticles (d_{av} = 10 nm, 0.01% by cement mass) and Al₂O₃ nanoparticles (d_{av} = 50 nm, 0.1% by cement mass). The kinetics of early structure formation were evaluated by plastic strength using a cone plastometer. Aggregative stability was determined by compression rheometry using an Instron 5982 testing machine. The phase composition was established by X-ray diffractometry. The microstructure of the specimens was studied by scanning electron microscopy. Compressive strength was recorded after 1 and 28 days. **Results.** It was found that Al₂O₃ nanoparticles most significantly accelerate the process of early coagulation-crystallization structure formation in cementitious materials (the exponent of the kinetic equation is 1.13 without a change in the controlling process), increase aggregative stability by a factor of 2–3, but reduce plastic deformation. SiO₂ nanoparticles maximally accelerate the hydration process (degree of hydration 93% after 28 days of hardening) while forming low-basic hydrosilicates. The 28-day compressive strength of cementitious materials modified with SiO₂ and Al₂O₃ nanoparticles was 92.7 MPa and 96.4 MPa, respectively, which is significantly higher than that of the reference system (51.1 MPa). **Discussion.** The twofold increase in strength is attributed to different mechanisms: for SiO₂ nanoparticles – their high pozzolanic activity and accumulation of high-strength cementitious phases; for Al₂O₃ nanoparticles – an increase in the aggregative stability of the coagulation structure, which ensures the formation of a homogeneous crystallization structure without major defects. **Conclusion.** Complex modifiers based on SiO₂ and Al₂O₃ nanoparticles enable targeted control of both the hydration activity and structural homogeneity of cementitious materials, providing a strength increase of up to 1.9 times.

KEYWORDS: SiO₂ and Al₂O₃ nanoparticles, cementitious material, aggregative stability, structure formation, plastic strength, hydration, strength

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Влияние наномодификаторов SiO_2 и Al_2O_3 на агрегативную устойчивость, структурообразование и прочность цементных систем

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АННОТАЦИЯ

Введение. Применение наноразмерных модификаторов на основе SiO_2 и Al_2O_3 открывает новые возможности управления структурой цементных композитов, однако их влияние на агрегативную устойчивость коагуляционной структуры и прочностные характеристики изучено недостаточно. Цель работы – установить влияние комплексных наномодификаторов на взаимосвязь агрегативной устойчивости, кинетики раннего структурообразования и прочности цементных систем. **Материалы и методы.** Исследованы три системы на основе портландцемента ЦЕМ I 42,5 с суперпластификатором марки Sika®ViscoCrete® 20 HE (эталон), а также с добавками нано- SiO_2 ($d_{cp} = 10$ нм, 0,01% от массы цемента) и нано- Al_2O_3 ($d_{cp} = 50$ нм, 0,1% от массы цемента). Кинетику раннего структурообразования оценивали по пластической прочности (конический пластометр), агрегативную устойчивость определяли методом реометрии при сдавливании (INSTRON 5982), фазовый состав устанавливали методом рентгеновской дифрактометрии, микроструктуру образцов изучали методом сканирующей электронной микроскопии, прочность при сжатии фиксировали через 1 и 28 суток. **Результаты.** Установлено, что наномодификатор Al_2O_3 наиболее выражено ускоряет процесс раннего коагуляционно-кристаллизационного структурообразования цементной системы (показатель степени кинетического уравнения составляет 1,13 без смены определяющего процесса), повышает агрегативную устойчивость в 2–3 раза, однако снижает пластическое деформирование. Наномодификатор SiO_2 максимально ускоряет процесс гидратации ($C_r = 93\%$ через 28 сут твердения), при этом формирует низкоосновные гидросиликаты кальция. Прочность при сжатии через 28 суток для наномодифицированных цементных систем составила 92,7 и 96,4 МПа, соответственно, что существенно выше, чем в эталонной системе (51,1 МПа). **Обсуждение.** Двукратный прирост прочности обусловлен различными механизмами действия наномодификаторов: для SiO_2 – их высокой пуццолановой активностью и накоплением высокопрочных цементирующих фаз, для Al_2O_3 – повышением агрегативной устойчивости коагуляционной структуры, что обеспечивает формирование однородной кристаллизационной структуры без крупных дефектов. **Заключение.** Комплексные наномодификаторы SiO_2 и Al_2O_3 позволяют целенаправленно регулировать как гидратационную активность, так и структурную однородность цементных систем, обеспечивая повышение прочности до 1,9 раза.

КЛЮЧЕВЫЕ СЛОВА: наномодификаторы, SiO_2 , Al_2O_3 , цементные системы, агрегативная устойчивость, структурообразование, пластическая прочность, гидратация цемента, прочность

ИСТОЧНИКИ ФИНАНСИРОВАНИЯ НАУЧНОЙ РАБОТЫ, РЕЗУЛЬТАТОМ КОТОРОЙ СТАЛА ПУБЛИКАЦИЯ: Работа выполнена при поддержке Министерства науки и высшего образования Российской Федерации в рамках темы государственного задания (проект FZGM-2026-0003).

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INTRODUCTION

The industrial implementation of nanotechnology has provided qualitatively new opportunities for controlling the structure and properties not only of nanostructures [1–6] but also of macroscale objects such as construction composites, which also possess nanoscale structural levels [3–16]. The literature analysis allows several key research

directions regarding the effect of nanoparticles on the concrete properties to be identified: controlling the mechanisms of hydration and structure formation [1–3], developing methods for the synthesis and dispersion of nanoparticles [4, 17–20], and creating materials with unique functional properties (self-sensing, self-healing) [21–25].

The most versatile effects in property regulation are achieved using SiO_2 and Al_2O_3 nanoparticles, which is

logically determined by their crystallochemical similarity to the phase-forming components of cementitious materials [1]. Studies [1, 7–16] propose a mechanism for the influence of these nanomodifiers on the hydration process. It has been shown that SiO_2 nanoparticles exhibit high pozzolanic activity and a pronounced nucleation effect. Optimal dosages in most cementitious materials lie in the range of 0.5–3% by weight of cement. Al_2O_3 nanoparticles, at dosages of 0.5–2% by weight of cement, are particularly effective in enhancing early-age strength and forming an aluminum-enriched C–S–H layer [8–10, 14–16]. These effects are associated with the dissolution of Al_2O_3 and the accelerated formation of the ettringite. Combined nanoadditives based on SiO_2 together with Al_2O_3 make it possible to combine pozzolanic activity, the nucleation effect, and specific chemical effects, ensuring high early and long-term strength, reduced water absorption, and increased corrosion resistance [7].

A less investigated issue is the effect of nanoadditives on the spatial-geometric modification of the structure, the control of its homogeneity characteristics, and, ultimately, on the mechanical properties of the material.

The significance of this problem for the structure formation of cementitious materials under nanomodification is determined by the following. The development of strength in cementitious materials is governed by the transition from a coagulation structure with reversible strength-related contacts to a crystallization structure with strong, irreversible intergrowth and overgrowth contacts. This process begins with the formation of a spatial structural network, which undergoes rapid strengthening as the concentration and dispersity of the solid phase increase during the hydration of clinker minerals. As a result, all types of defects and inhomogeneities of the coagulation structure are “inherited” during the formation of the crystallization structure and become irreversibly fixed within it, leading to a reduction in the material’s performance properties. Consequently, to enhance the strength of cementitious materials, it is necessary to create, already at the early stages of structure formation, the prerequisites for forming a maximally homogeneous structure free from major defects. Of primary importance here is the aggregative stability of the coagulation structure of cementitious materials – its ability to preserve the spatial structural network under external mechanical actions that are inevitable

in manufacturing processes (e.g., during transportation, placement, etc.).

This article is devoted to the study of the influence of complex modifiers based on SiO_2 and Al_2O_3 nanoparticles on the relationship between the aggregative stability of the coagulation structure and the characteristics of the crystallization structure and strength of cementitious materials.

The research involved:

- Establishing the effect of complex nanomodifiers on the early-age structure formation kinetics of cementitious materials.
- Evaluating the effect of complex nanomodifiers on the aggregative stability of the coagulation structure of viscoplastic cementitious materials under mechanical loading.
- Determining the effect of complex nanomodifiers on the phase composition, hydrate phase structure, and strength of the crystallization structure of hardened cementitious systems.

MATERIALS AND METHODS

Three types of cementitious materials were studied, as presented in Table 1.

The following raw components were used in the systems: Portland cement (C) CEM I 42.5 manufactured by “EUROCEMENT GROUP” (Russian State Standard GOST 31108–2016); a polycarboxylate ether-based superplasticizer (SP) of the brand Sika®ViscoCrete® 20 HE; and tap water (W) (Russian State Standard GOST 23732–2011). Two types of complex additives based on nanoparticles (NP) were used for modification: (1) NP SiO_2 ($d_{av} \sim 10$ nm) + SP, and (2) NP Al_2O_3 ($d_{av} \sim 50$ nm) + SP. The nanoparticles were synthesized using the sol-gel method, which is described in detail in [17–20]. The superplasticizer was used to stabilize the resulting nanosized sols. The water-cement ratio (W/C) for each system was selected to maintain a normal consistency of the cement paste.

The mixtures were prepared in a “MATEST-E095” mixer by mixing the dry components with the solution for 3 minutes.

The kinetics of early-age structure formation were assessed by measuring the plastic strength using a self-developed cone plastometer (Fig. 1).

Table 1. Mixture proportioning of the cementitious systems

No.	Components	System ID	Additive proportioning		W/C
			SP, % solution volume	NP, % mass cement	
1	Cement, water, superplasticizer (reference)	C–W–SP	0.2	–	0.33
2	Cement, water, superplasticizer, SiO_2	C–W–SP– SiO_2	0.2	0.01	0.27
3	Cement, water, superplasticizer, Al_2O_3	C–W–SP– Al_2O_3	0.2	0.1	0.30



Fig. 1. Self-developed cone plastometer

The operating principle of the plastometer is based on determining the plastic strength from the load required to penetrate a metal cone with a given apex angle into the test material to a specified depth. For the experimental investigations, the cementitious mixture was placed immediately after mixing into 10 molds ($a = 100$ mm, $b = 150$ mm, $h = 60$ mm). In each mold, one measurement was taken between the time of placement and 100 minutes at 10-minute intervals.

Based on the data obtained, kinetic curves of plastic strength $P_{pl} = f(\tau)$ were constructed. The experimental results were processed as follows. The primary experimental kinetic curves of plastic strength were described formally using the kinetic equation:

$$P_{pl} = (k \cdot \tau)^n, \quad (1)$$

where P_{pl} is the plastic strength (kPa), τ is the setting time (min), k is the rate constant of hydration process ($\text{kPa}^{1/n} \cdot \text{min}^{-1}$), and n is the exponent of the equation, reflecting the change in the rate-determining hydration process.

The logarithm of Equation (1) was taken to obtain an equation of the form:

$$\ln(P_{pl}) = n \cdot \ln(k) + n \cdot \ln(\tau). \quad (2)$$

The plastic strength as a function of time was plotted in double logarithmic coordinates. The n -value for each system studied was calculated using the least squares method. The early-stage cement hydration process is complex, involving multiple parallel and coupled reactions. Therefore, it is useful to introduce the terms “rate-determining process” and “rate-controlling process”. The rate-determining process is the parallel process that contributes most to structure formation and, consequently, to the increase in plastic strength at a given stage. This process may consist of several sequential steps, the slowest of which is the rate-controlling process. When the kinetic exponent $n > 1$, the rate-determining process is the chemical hydration reaction. When $n < 1$, the kinetics are controlled by diffusion of components into the reaction zone.

The aggregative stability of cementitious systems in the viscoplastic state was studied using squeeze rheometry with an INSTRON 5982 testing system. Immediately after preparation of the cementitious mixture, cylindrical specimens with dimensions $R = h_0 = 25$ mm were fabricated. These were placed between two smooth plates of the INSTRON 5982 system, the diameter of which matched the specimen size. The tests were conducted at a constant loading rate of 0.5 N/s. During the experiment, load–displacement (P – Δ) curves were recorded. At the inflection points of the curves, corresponding to (1) the onset of deformation and (2) the onset of loss of continuity, the values of structural strength and plastic strength were determined using the relationship:

$$\sigma = \frac{P}{\pi R^2}, \text{ kPa}. \quad (3)$$

These values were adopted as criteria for the aggregative stability of the cementitious systems. The structural strength σ_0 , determined at the first inflection point of the P – Δ curve, characterizes the ability of the viscoplastic mixture to withstand load without deforming. The plastic strength σ_{pl} characterizes the ability of the viscoplastic mixture to withstand load and deform without loss of continuity. As an additional criterion, the relative deformation of the mixture layer up to the point of loss of continuity, Δ_{pl} (mm/mm), was evaluated.

The compressive strength of hardened cement pastes was evaluated by testing specimens of size $5 \times 5 \times 5$ cm after 1 and 28 days of curing under normal conditions ($t = 20 \pm 2$ °C, $RH = 100\%$) using an Instron Sates 1500HDS testing machine. Each series consisted of six specimens, and the within-series coefficient of variation of the test results did not exceed 7%.

The following physicochemical methods were used to investigate the structure of the cementitious systems. The qualitative composition of the synthesized nanoadditives,

as well as the phase composition of the nanomodified cement stone, was studied using powder X-ray diffraction (ARL X'TRA diffractometer, $\text{CuK}\alpha$ radiation, $\lambda = 1.541788 \text{ \AA}$). The experimental data were processed using the PDWin 4.0 software package [26]. The degree of hydration of the cementitious systems was calculated from the content of the alite phase ($3\text{CaO}\cdot\text{SiO}_2$ (C_3S)) by comparing their diffractograms with that of a pure cement clinker sample [27]:

$$D_h(\text{C}_3\text{S}) = (1 - I_{\text{mod}}/I_0) \times 100\%, \quad (4)$$

where I_{mod} is the intensity of the diffraction maximum at $d = 2.75 \text{ \AA}$ of the C_3S phase for samples of different compositions (by additive type) and different cement hydration times, and I_0 is the intensity of the diffraction maximum at $d = 2.75 \text{ \AA}$ of the C_3S phase of the original cement.

The microstructure of the cementitious systems was characterized by scanning electron microscopy (SEM, Phenom XL, accelerating voltage = 15 kV, pressure = 0.10 Pa) using fractured specimens of hardened cement paste.

RESULTS

Early Structure Formation Kinetics

The experimental curves of plastic strength (P_{pl}) evolution over time for the cementitious systems (Fig. 2a) exhibit similar patterns and have three typical stages:

1. Change in P_{pl} at a constant rate;
2. A plateau where the increase in P_{pl} slows down;
3. An acceleration of the increase in P_{pl} .

The time for each stage depends on the type of nanoparticle. The most intensive increase in plastic strength is characteristic of the $\text{C-W-SP-Al}_2\text{O}_3$ system, while the slowest increase is observed for the C-W-SP-SiO_2 system. This is determined by the nature of the rate-determining process of early structure formation and its rate-controlling stage, as evidenced by the n -values of the plastic kinetic equations for plastic strength (Fig. 2b).

For the C-W-SP system, the n -values of the plastic kinetic equation is 1.05 up to 50 minutes, indicating a chemical reaction. After 60 minutes, the process becomes controlled by diffusion of components into the reaction zone, with $n = 0.60$.

For the C-W-SP-SiO_2 system, the n -values of the plastic kinetic equation is 1.09 up to 60 minutes, indicating a chemical reaction. After 70 minutes, the process becomes controlled by diffusion of components into the reaction zone, with $n = 0.43$.

For the $\text{C-W-SP-Al}_2\text{O}_3$ system, the n -values of the plastic kinetic equation remains unchanged throughout the entire test duration (100 min) and corresponds to a chemical reaction, with $n = 1.13$.

Investigation of Aggregative Stability

The squeezing tests produced two types of curves: «relative displacement Δ – time τ » (Fig. 3a) and «load P – relative displacement Δ » (Fig. 3b). Three main zones can be identified on the $\Delta - t$ curves:

- «Placing phase», the structure maintains stability;
- «Perfect plastic response phase», viscoplastic fluid without damaged structure;
- «Plastic fluid» with damaged structure.

On the $P - \Delta$ curves, load drop peaks are observed, which correspond to the onset of loss of structural continuity, i.e., the beginning of structure failure. These points correspond to the beginning of the structure failure zone on the $\Delta - t$ curves.

Interpretation of the obtained curves shows that the C-W-SP and C-W-SP-SiO_2 systems have lower values of structural strength σ_0 and plastic strength σ_{pl} , as well as larger plastic deformations Δ_{pl} , compared to the $\text{C-W-SP-Al}_2\text{O}_3$ system (Table 2).

Based on the squeezing tests, the influence of complex nanomodifiers on the aggregative stability of cementitious systems under load was established. The introduction of nanomodifiers increases aggregative stability. For these systems, the values of structural strength σ_0 , which characterize the ability of the viscoplastic mixture to withstand load without deforming, are 2–3 times higher than those of the unmodified system. The $\text{C-W-SP-Al}_2\text{O}_3$ system exhibits the highest aggregative stability but shows a reduced ability to deform without loss of structural continuity. The use of the SiO_2 -based nanomodifier has practically no effect on the plastic deformation capacity.

Microstructural Characteristics and Strength

According to XRD data, the phase composition of the investigated cementitious systems is characterized by the following features (Fig. 4).

For the C-W-SP system, the main hydrate phases are tobermorite like phase ($4\text{CaO} \cdot 5\text{SiO}_2 \cdot 5\text{H}_2\text{O}$) and C-S-H (I) phase of variable composition ($x\text{CaO} \cdot \text{SiO}_2 \cdot z\text{H}_2\text{O}$). Small amounts of high basic hydrosilicate phases ($2\text{CaO} \cdot \text{SiO}_2 \cdot \text{H}_2\text{O}$) and Ca(OH)_2 are also detected.

For the C-W-SP-SiO_2 system, low basic calcium hydrosilicate phases ($x\text{CaO} \cdot \text{SiO}_2 \cdot z\text{H}_2\text{O}$) and high basic calcium hydrosilicate phases ($2\text{CaO} \cdot \text{SiO}_2 \cdot 0.5\text{H}_2\text{O}$, $2\text{CaO} \cdot \text{SiO}_2 \cdot \text{H}_2\text{O}$) are observed. The portlandite phase (Ca(OH)_2) is absent. Apparently, a reaction occurs between Ca(OH)_2 and the active SiO_2 nanoparticles, forming calcium hydrosilicates. This is consistent with reports by other authors [7, 9, 13, 14].

The $\text{C-W-SP-Al}_2\text{O}_3$ system exhibits the greatest diversity of hydrate phases. This suggests that Al_2O_3 , being thermodynamically more stable, reacts more slowly, creating concentration gradients and allowing numerous

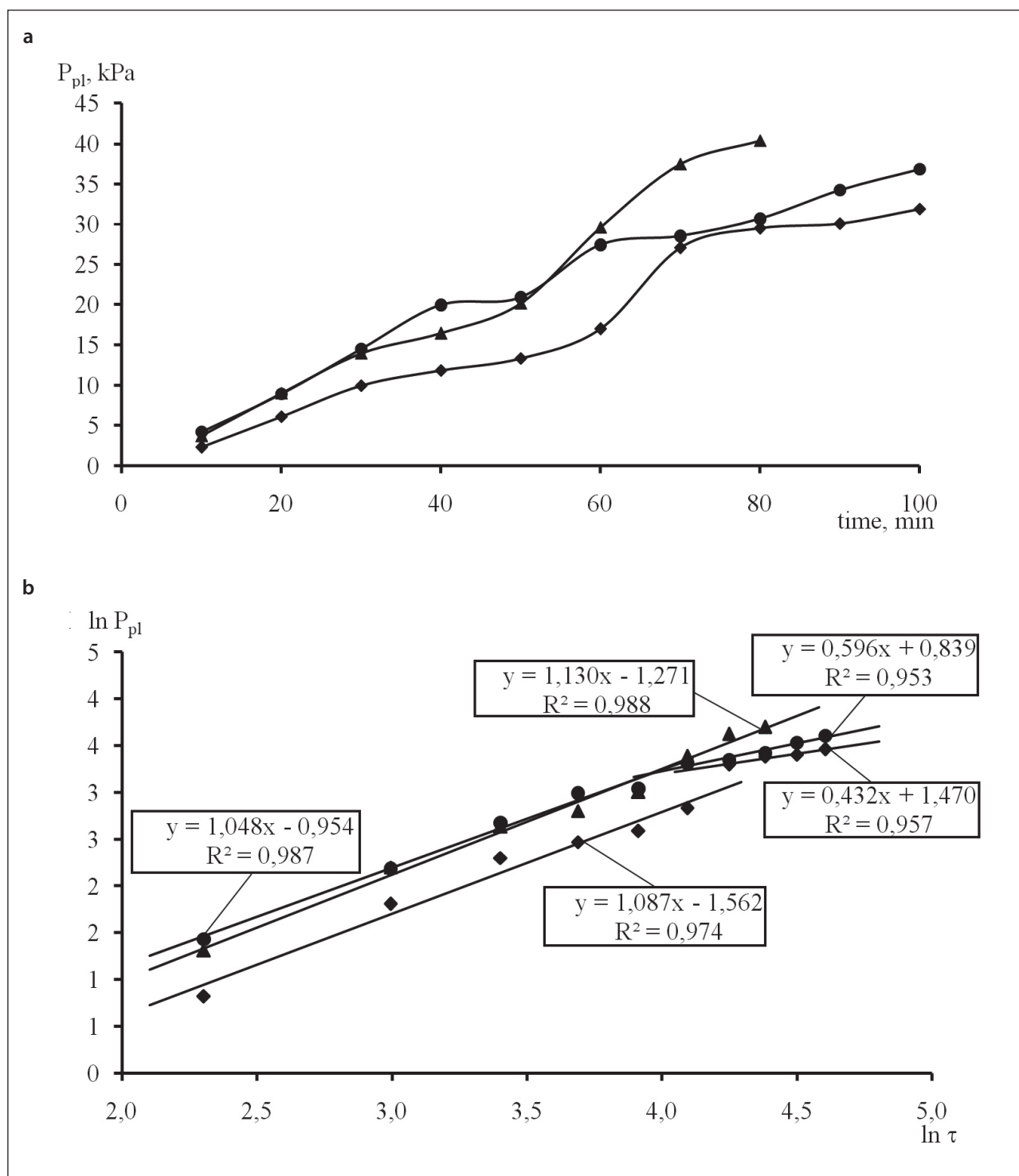


Fig. 2. Kinetics of early structure formation of cementitious systems: (a) experimental kinetic plastic strength curves; (b) plastic strength kinetic curves in logarithmic coordinates.

Legend: • – C–W–SP; ◆ – C–W–SP–SiO₂; ▲ – C–W–SP–Al₂O₃.

intermediate and metastable phases (calcium hydroaluminates of varying basicity) to crystallize. The main newly formed phases in the cement stone are Al(OH)₃, CaO · Al₂O₃ · 2SiO₂ · 4H₂O, 5CaO · 2SiO₂ · H₂O, and

(CaO)_x · SiO₂ · zH₂O. Small amounts of CaO · Al₂O₃ · 8.5H₂O and CaO · Al₂O₃ · 6SiO₂ · 6H₂O are also detected.

SEM analysis revealed the following microstructural features of the cementitious materials (Fig. 5):

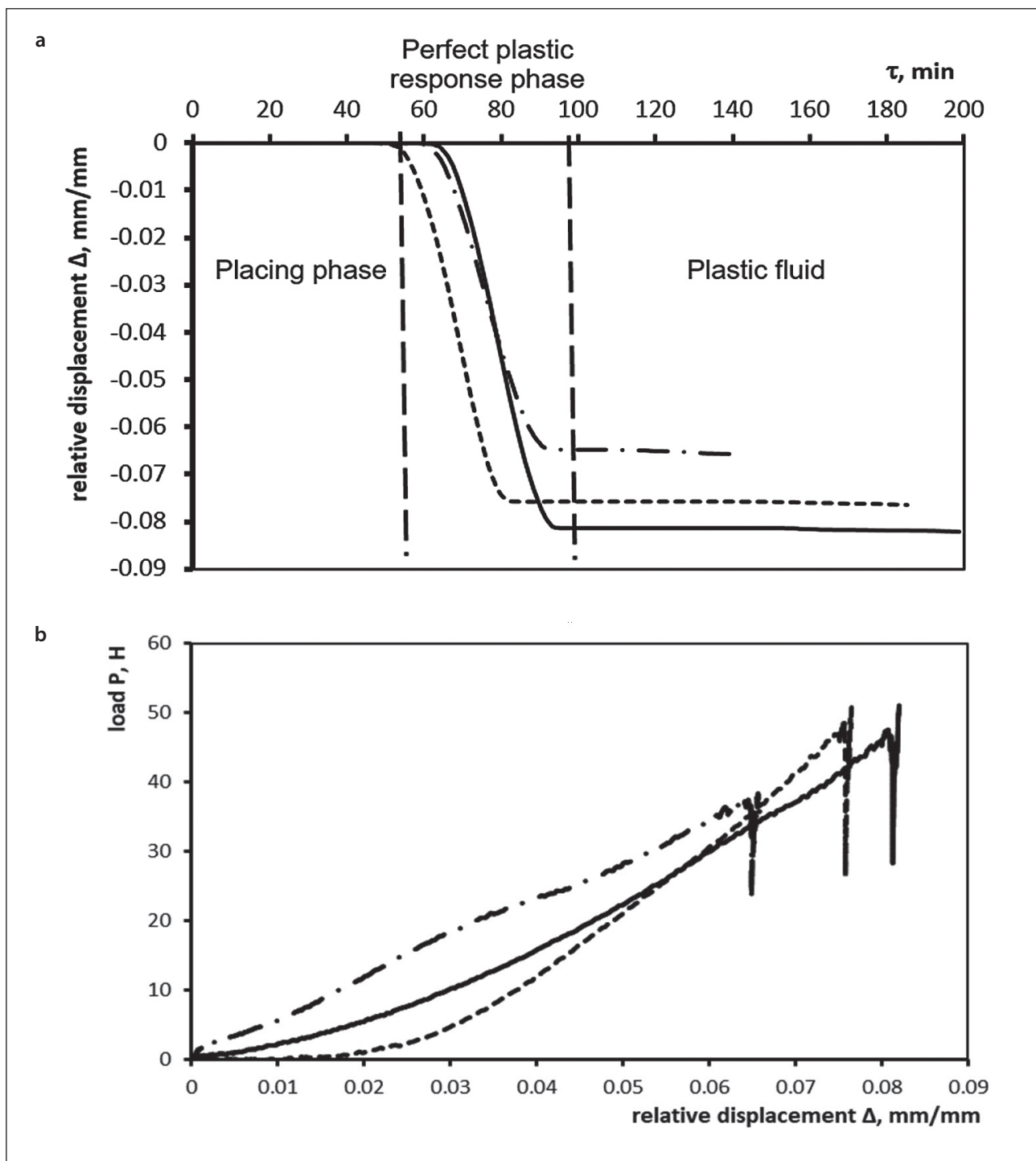


Fig. 3. Tested cement pastes experimental results fresh cement paste: (a) «relative displacement Δ – time τ »; (b) «load P – relative displacement Δ ».

Legend: --- C–W–SP; — C–W–SP–SiO₂; - · - C–W–SP–Al₂O₃

1. In the reference C–W–SP system, a homogeneous structure with an amorphous-crystalline appearance is formed.

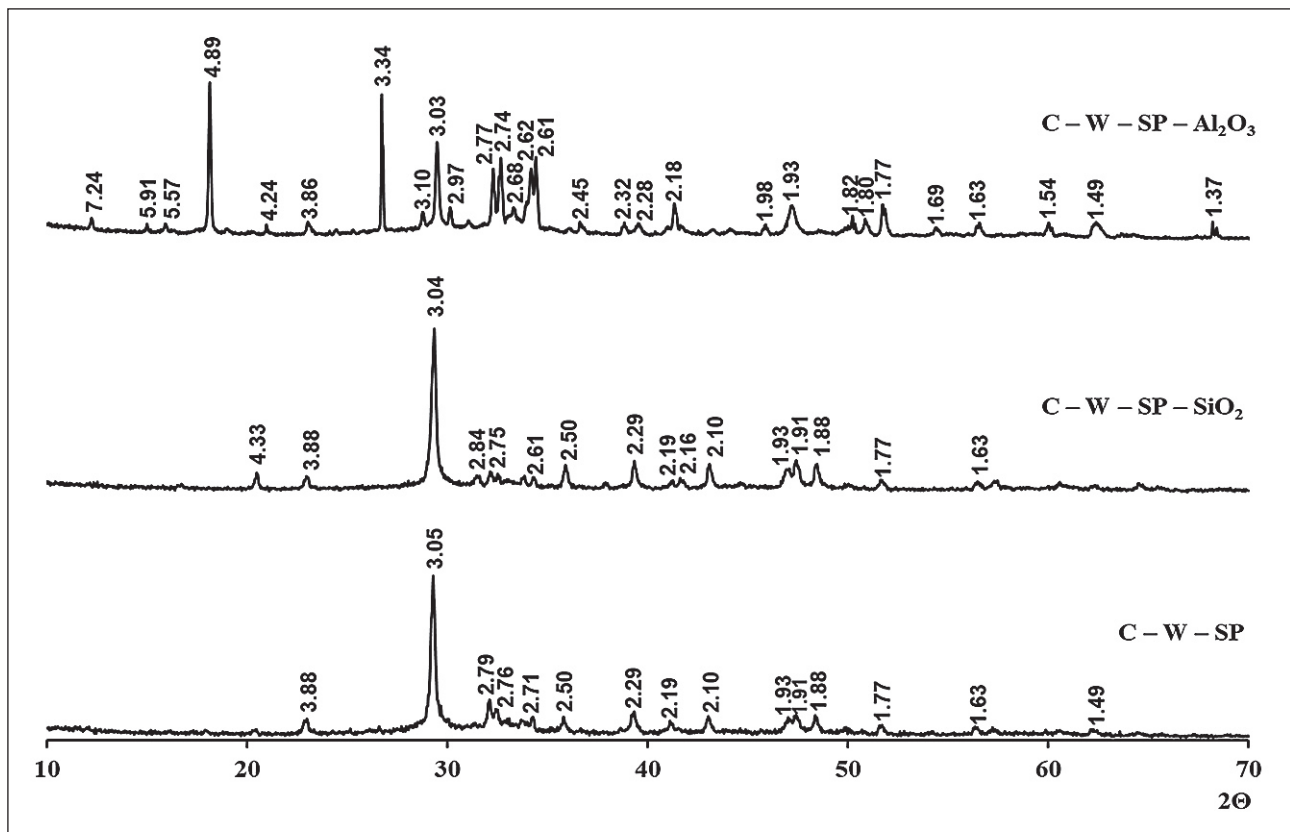
2. In the C–W–SP–SiO₂ system, an amorphous-crystalline structure is also formed, but it is denser and

more crystallized, with inclusions of small, platy C–S–H crystals.

3. In the C–W–SP–Al₂O₃ system, a highly homogeneous, dense, and predominantly amorphous structure is formed.

Table 2. Aggregative stability criteria of the cementitious materials

System ID	Structural strength σ_0 , kpa	Plastic strength σ_{pl} , kpa	Relative deformation, Δ_{pl} , mm/mm
C–W–SP	0.9	48.5	0.075
C–W–SP–SiO ₂	1.7	47.5	0.080
C–W–SP–Al ₂ O ₃	2.1	37.8	0.065

**Fig. 4.** X-ray diffractogram patterns of cementitious materials.

Legend: $x\text{CaO} \cdot \text{SiO}_2 \cdot z\text{H}_2\text{O}$ ($d = 5.60, 3.07, 2.97, 2.80, 2.28, 2.15$); $4\text{CaO} \cdot 5\text{SiO}_2 \cdot 5\text{H}_2\text{O}$ ($d = 4.22, 3.14, 3.05, 2.11, 1.92, 1.84$); $2\text{CaO} \cdot \text{SiO}_2 \cdot 0.5\text{H}_2\text{O}$ ($d = 2.87, 2.77, 2.67, 2.50, 2.58, 1.98, 1.81, 1.62$); $2\text{CaO} \cdot \text{SiO}_2 \cdot \text{H}_2\text{O}$ ($d = 4.32, 3.82, 3.04, 2.84, 2.70, 1.90, 1.80, 1.66$); $\text{CaO} \cdot \text{Al}_2\text{O}_3 \cdot 2\text{SiO}_2 \cdot 4\text{H}_2\text{O}$ ($d = 8.97, 7.90, 6.61, 5.10, 3.97, 3.90$); $5\text{CaO} \cdot 2\text{SiO}_2 \cdot \text{H}_2\text{O}$ ($d = 5.42, 3.98, 3.79, 3.31, 3.02, 2.92$); $\text{Al}(\text{OH})_3$ ($d = 4.85, 4.37, 4.32, 2.45, 2.42, 2.39$)

The complex nanoadditives investigated significantly accelerated the hydration and hardening processes (Table 3). The compressive strength of the nanoparticle-modified cementitious materials was 2.5–3 times higher after 1 day of curing and 1.8–1.9 times higher after 28 days compared to the reference system without nanoparticles.

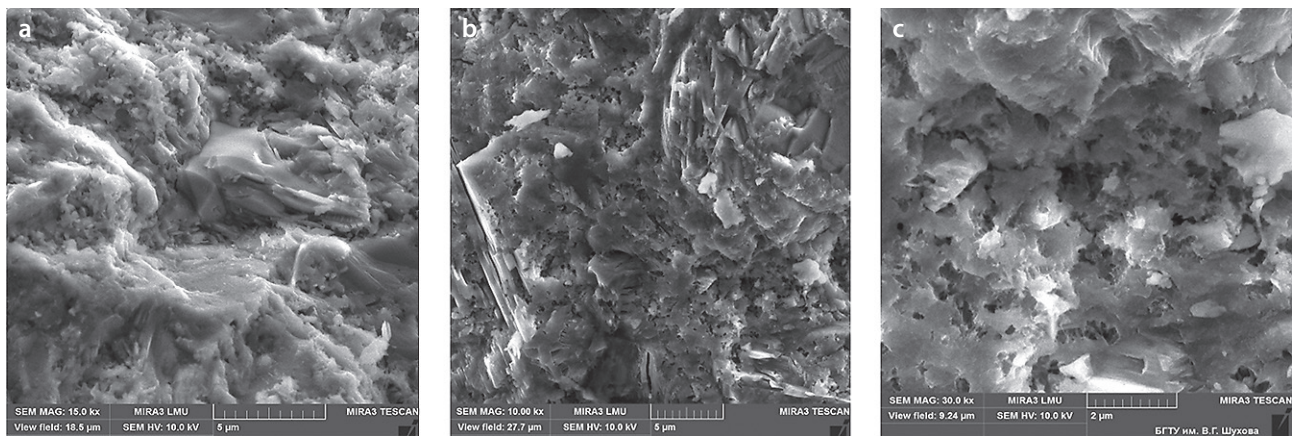
DISCUSSION

Kinetic of the early structure formation (setting) process showed that the effectiveness of the additives is determined by differences in the mechanisms of the initial

stages of structure formation (see Fig. 2). For the system containing the Al₂O₃-based nanomodifier, the n -values of the plastic kinetic equation is higher than for the C–W–SP and C–W–SP–SiO₂ systems. The value $n = 1.13$ corresponds to a chemical reaction (see Table 3). Throughout the entire setting time (0–100 min), no change in the rate-determining process is observed. For the C–W–SP and C–W–SP–SiO₂ systems, the rate-determining process also corresponds to a chemical reaction during the period from 0 to 50 min. After 50 min, a change occurs from a chemically controlled to a diffusion-controlled process, which is governed by the rate of water diffusion to the cement grains. Thus, in the presence of Al₂O₃ par-

Table 3. Characteristics of structure formation and hardening of the cementitious systems

System ID	n-values of the plastic kinetic equation		Degree of hydration, % (28 days)	Compressive strength, R_{c} , MPa	
	$\tau_{0-50 \text{ min}}$	$\tau_{60-100 \text{ min}}$		1 day	28 days
C–W–SP–	1.05	0.60	62	20.3	51.1
C–W–SP–SiO ₂	1.09	0.43	93	64.1	9.7
C–W–SP–Al ₂ O ₃	1.13		68	51.1	96.4

**Fig. 5.** SEM-photo of the cementitious materials after 28 days hardening : a) C–W–SP–; b) C–W–SP–SiO₂; c) C–W–SP–Al₂O₃

ticles, the early stage of coagulation–crystallization structure formation is accelerated, which may be due to the removal of diffusion barriers and an increase in the active interfacial area between the reacting phases.

Based on the obtained data, it can be stated that the introduction of nanomodifiers enhances the aggregative stability of the primary coagulation structure of the cementitious systems. At the same time, the Al₂O₃-based nanomodifier accelerates the formation of the crystallization structure during the early structure formation stage.

The most significant effect of the SiO₂-based nanomodifier is that, due to the increase in the active interfacial area at the very beginning of the structure formation process, the rate of cement hydration increases sharply. Consequently, the high strength values of the hardened cement paste are directly related to the realization of the hydration potential of the cement minerals and the accumulation of hydrate phases. The increase in strength is also influenced by the presence of predominantly low-basic calcium hydrosilicate ($x\text{CaO} \cdot \text{SiO}_2 \cdot z\text{H}_2\text{O}$) in the structure and the absence of the low-strength portlandite phase $\text{Ca}(\text{OH})_2$.

The effect of the Al₂O₃-based nanomodifier on the strength development of cementitious systems is associated with an increase in the strength and stability of the primary coagulation structure, which promotes a more

homogeneous distribution of hydrate phases within the crystallization structure (see Fig. 5). As a result, despite comparable degrees of cement hydration in the C–W–SP ($D_h = 62\%$) and C–W–SP–Al₂O₃ ($D_h = 68\%$) systems, the strength of the Al₂O₃-modified system is 1.9 times higher than that of the unmodified system. The positive contribution to strength of the aluminates and aluminosilicate phases detected in the cement stone structure should also not be excluded.

CONCLUSION

The introduction of complex nanomodifiers accelerates the early coagulation–crystallization structure formation of cementitious systems, which may be attributed to the removal of diffusion barriers and an increase in the active interfacial area between the reacting phases. This effect is particularly pronounced in the presence of Al₂O₃ particles.

Nanomodified cementitious systems exhibit enhanced aggregative stability under loading. For these systems, the structural strength σ_0 , which reflects the load a viscoplastic mixture can bear without deforming, is 2–3 times higher than that of the unmodified system.

The nearly twofold increase in strength of the nanomodified cementitious systems is associated with two

factors. In the case of SiO_2 nanoparticles, this is due to the preferential formation and increased content of high-strength, low-basic hydrate phases. The introduction of Al_2O_3 nanoparticles does not significantly affect the degree of cement hydration but enhances the aggregative stability of the coagulation structure of the cementitious

system and its ability to preserve the spatial structural network under external mechanical loading. As a result, already at the early stages of structure formation, the prerequisites are created for forming a maximally homogeneous structure free from major defects, which ultimately determines its strength.

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ADDITIONAL INFORMATION

The authors declare that generative artificial intelligence technologies and technologies based on artificial intelligence were not used in the preparation of the article.

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O.V. Artamonova – selection and analysis of scientific literature; analysis of experimental data; writing the results and discussion section of the article.

E.A. Britvina – conducting experimental studies; analysis of experimental data; preparation of figures for the article.

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